

## LETTERS TO THE EDITOR

# Synthesis and Intramolecular Heterocyclization of *N*-Anabasinylacetic Acid *N*-Phenylthiosemicarbazide

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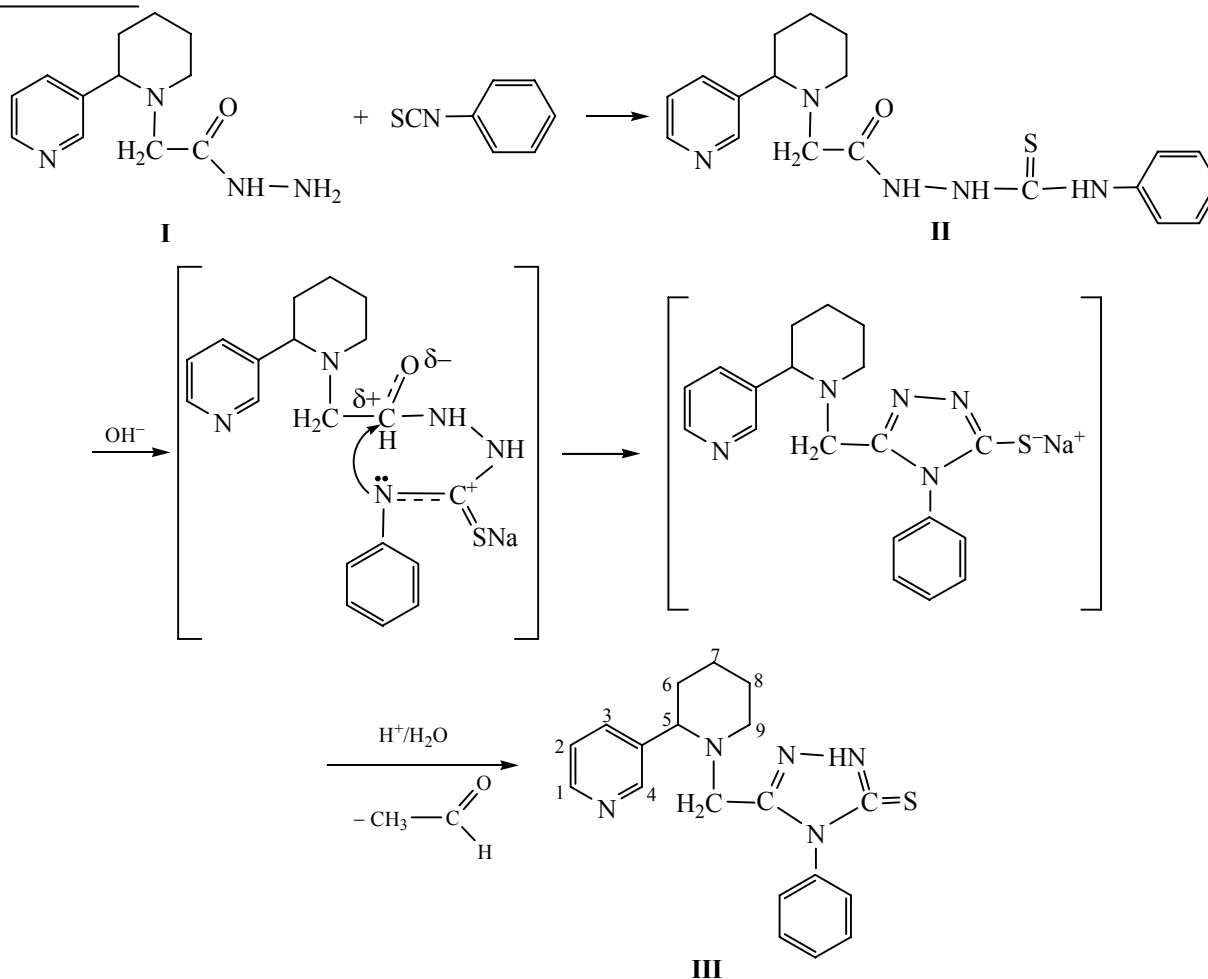
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One of the promising directions in screening of the new potentially antibacterial derivatives of alkaloid anabasine is preparation of its thiosemicarbazide derivatives, since they are known for a wide range of biological activity [1].

Extending the study on modification of anabasine [2], we synthesized *N*-anabasinylacetic acid *N*-phenylthiosemicarbazide **II** by interaction of *N*-anabasinylacetic acid hydrazide **I** with phenylisothiocyanate in ethanol, having achieved the yield of 74%.



Acidification of aqueous alkaline solution of the prepared phenylthiocarbazide derivative **II** led to its intramolecular heterocyclization to form 4-phenyl-5-anabasinomethyl-1,2,4-triazole-3-thione **III**.

Compounds **II** and **III** were white crystalline solids, soluble in many polar and nonpolar organic solvents. Composition and structure of **II** and **III** were confirmed by elemental analysis, IR and  $^1\text{H}$  NMR spectroscopy.

The initial *N*-anabasinylacetic acid hydrazide **I** was obtained as described in [3].

***N*-Anabasinylacetic acid *N*-phenylthiosemicarbazide (II).** 1.35 g (0.011 mol) of phenylisothiocyanate was added to a solution of 2.34 g (0.01 mol) of the *N*-anabasinylacetic acid hydrazide in 10 ml of 2-propanol. The mixture was refluxed at 50–60°C for 10 h. The reaction progress was monitored by TLC. After cooling down, the appeared white crystalline precipitate was filtered off, washed with small amount of cold 2-propanol, and recrystallized from 2-propanol. Yield 2.73 g (74%), mp 100–102°C. IR spectrum,  $\nu$ ,  $\text{cm}^{-1}$ : 3279, 3272, 3106 (NH), 1660 (C=O), 1510–1590 (NH-Ar), 1270 (C=S).  $^1\text{H}$  NMR,  $\delta$ , ppm ( $J$ , Hz): 1.25–1.70 m (6H,  $\text{H}^6$ ,  $\text{H}^7$ ,  $\text{H}^8$ ), 2.96 t (1H,  $\text{H}^5$ ,  $J$  14.04), 8.36 d.d (1H,  $\text{H}^2$ ,  $J$  1.68), 8.60 s (1H,  $\text{H}^4$ ), 7.46 d (1H,  $\text{H}^3$ ,  $J$  7.8), 8.51 d.d (1H,  $\text{H}^1$ ,  $J$  1.65), 7.33 s (2H,  $\text{CH}_2$ ), 10.40 s [1H,  $\text{NH}(\text{C}=\text{O})$ ], 9.80 s (1H, NH), 13.72 s [1H,  $\text{NH}(\text{C}=\text{S})$ ], 7.00–7.80 m (5H, Ar). Found, %: C 61.76; H 6.27; N 18.95.  $\text{C}_{19}\text{H}_{23}\text{N}_5\text{OS}$ . Calculated, %: C 61.89; H 6.31; N 18.99.

**4-Phenyl-5-anabasinomethyl-1,2,4-triazole-3-thione (III).** 3.69 g (0.01 mol) of *N*-anabasinylacetic acid *N*-phenylthiosemicarbazide **II** was added to the aqueous

alkaline solution of 0.40 g (0.01 mol) of NaOH in 30 ml of distilled water. The reaction mixture was heated at 85°C for 2 h. After cooling, the mixture was neutralized with hydrochloric acid to pH 6–7. The formed precipitate was filtered off and recrystallized from 2-propanol. Yield 1.40 g (40%), milky powder, mp 135–136°C. IR spectrum,  $\nu$ ,  $\text{cm}^{-1}$ : 1619 (C=N), 1270 (C=S).  $^1\text{H}$  NMR spectrum (500 MHz),  $\delta$ , ppm ( $J$ , Hz): 1.20–1.65 m (6H,  $\text{H}^6$ ,  $\text{H}^7$ ,  $\text{H}^8$ ), 2.96 t (1H,  $\text{H}^5$ ,  $J$  14.04), 7.07 d.d (1H,  $\text{H}^2$ ,  $J$  4.78), 8.14 d.d (1H,  $\text{H}^4$ ,  $J$  1.86), 7.22 d (1H,  $\text{H}^3$ ,  $J$  7.28), 8.36 d.d (1H,  $\text{H}^1$ ,  $J$  4.73), 7.36 s (2H,  $\text{CH}_2$ ), 13.73 s (1H, NH), 7.54–7.62 m (5H, Ar). Found, %: C 64.93; H 6.02; N 19.93.  $\text{C}_{19}\text{H}_{21}\text{N}_5\text{S}$ . Calculated, %: C 65.09; H 6.29; N 20.11.

IR spectra were recorded with Nicolet AVATAR-320 Fourier-spectrometer in KBr pellets. The  $^1\text{H}$  NMR spectra were recorded with Bruker DRX500 spectrometer (500 MHz) in  $\text{DMSO}-d_6$ ; TMS was used as internal standard. TLC analysis was performed on Sorbfil plates using isopropyl alcohol–benzene–ammonia mixture (10:5:2) as eluent and detection with iodine vapors. Melting points were determined with Boetius apparatus.

## REFERENCES

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